

A glowing orange kimberlite pipe rises from a dark, textured ground. The ground is covered with numerous small, colorful mineral inclusions, including green, red, and white circular spots. The pipe itself is a bright, vertical column of orange light, tapering slightly towards the top where it glows most intensely. The overall scene is dramatic and highlights the geological features of a kimberlite pipe.

**New models for kimberlite
parental melts: composition,
temperature, ascent and
emplacement**

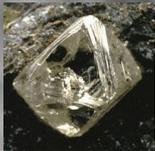
Kimberlites: a major source of diamonds



are exceptionally rare, but widespread on all major cratons;



originate in a diamond-stability field (> 40 kbar), but do not crystallise diamonds;



are fast ascending from mantle depths with a load of mantle and crustal xenoliths;



are emplaced in small-volume dykes and diatremes;



highly explosive emplacement and eruptions;



"...is regarded by ...the geological and geochemical community with an aura of glamour and mystique" (Eggler, 1989)

Kimberlite Parental Melt: existing dogma

Aphanitic Whole Rocks **==** **Parental Melt**

Definition: "Kimberlites are a clan of volatile-rich (dominantly CO_2) potassic ultrabasic rocks..." (R. H. Mitchell, 1986)

Hypabyssal group-I kimberlites are:

1. Ultramafic ($\text{MgO} > 22 \text{ wt\%}$; $\text{Mg\#} > 85$);
2. High-Fo olivine cumulate, rare other minerals, no liquidus pyroxenes;
3. Low "basaltic" component: high $\text{SiO}_2/\text{Al}_2\text{O}_3 > 8$, low $\text{Na}_2\text{O} < 0.2 \text{ wt\%}$;
4. Strong and variable hydration $> 2.7 \text{ wt\% H}_2\text{O}$ (serpentine), $\text{H}_2\text{O}/\text{CO}_2 > 1$;
5. Exceptional enrichment in incompatible trace elements and depletion in HREE



Unnamed kimberlite from Antarctica

Problem: "...the kimberlitic rock is both a **contaminated** and **altered** sample of its parent melt" (J. D. Pasteris, 1984)



What is wrong with an ultrabasic melt...?

Phenocrysts assemblage: olivine phenocrysts only, "melt"-olivine disequilibrium, no other common phenocrysts (pyroxenes & plagioclase);

Composition: mismatch between low-Al, low-Na kimberlite and other silicate magmas;

Composition: mismatch high Mg# (high-F) and trace/volatile elements enrichment (low-F);

Temperature: mismatch between calculated temperatures calculated ($>1350^{\circ}\text{C}$) and "recorded" temperatures ($<800^{\circ}\text{C}$);

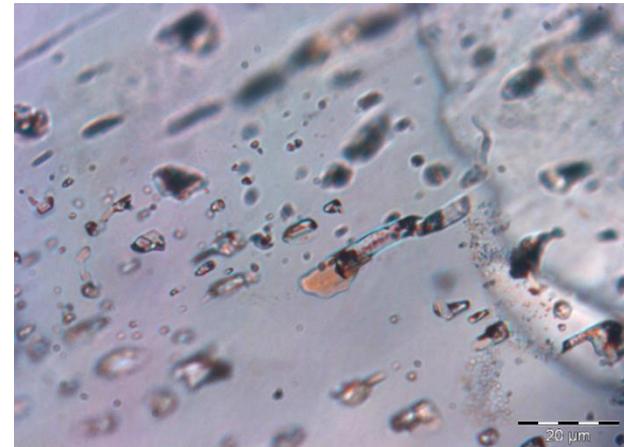
Rheology: mismatch between inferred mush of ultramafic melt + abundant solids (crystals/xenoliths) and low viscosity - low density - high buoyancy;

Unique explosivity style: what explodes in the ultramafic magma?

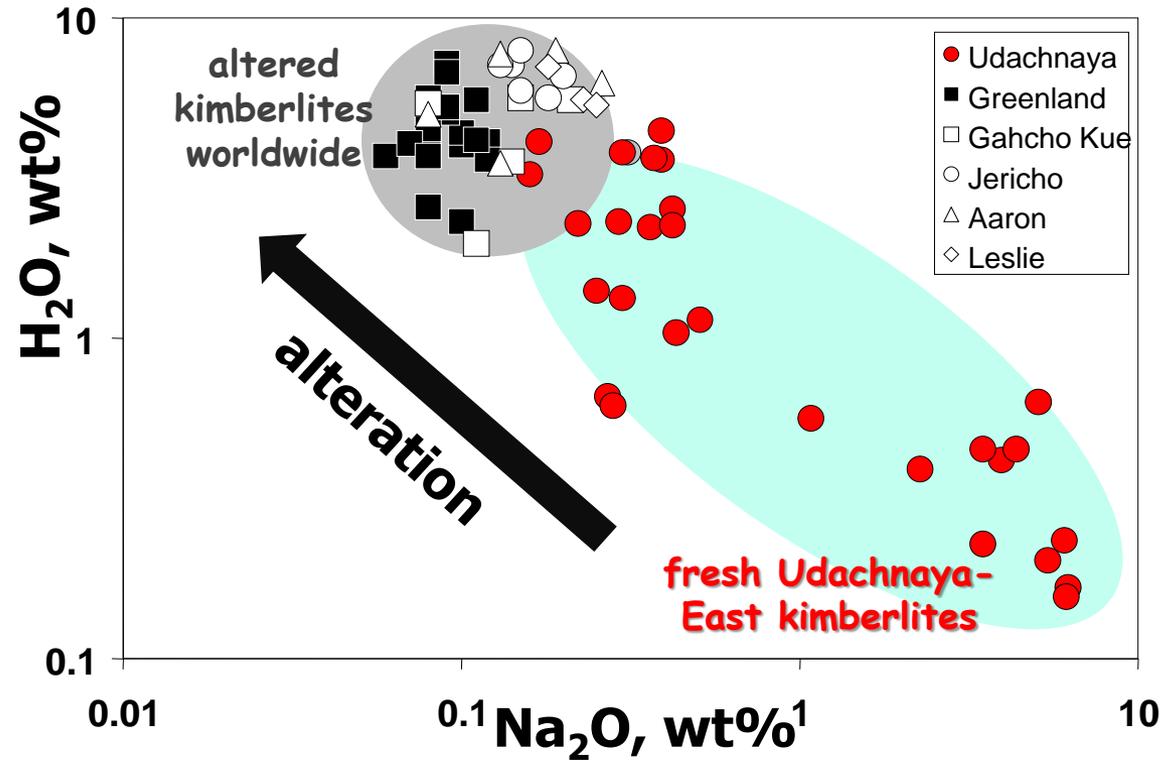
Our approach to kimberlite parental melt

- Study of least altered kimberlite samples;
- Study of groundmass assemblage;
- Study of olivine phenocrysts;
- Study of melt inclusions in olivine, Cr-spinel, perovskite, apatite, monticellite, phlogopite etc

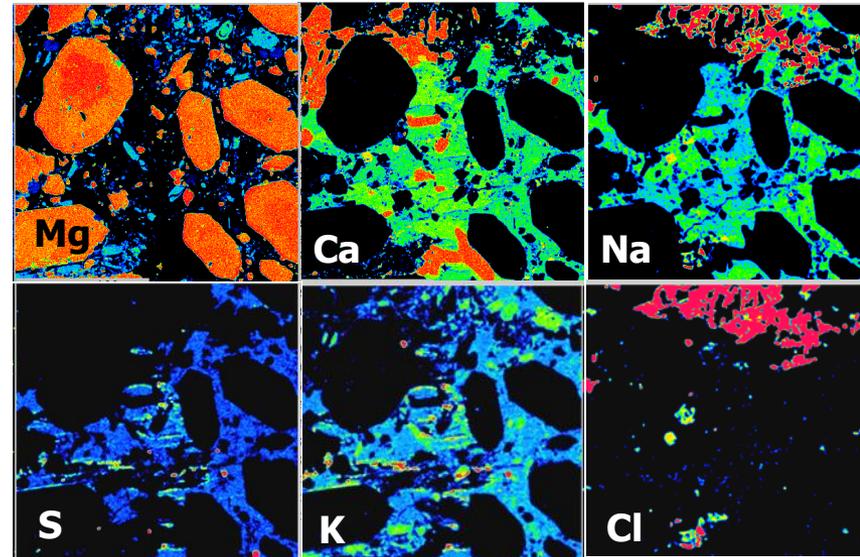
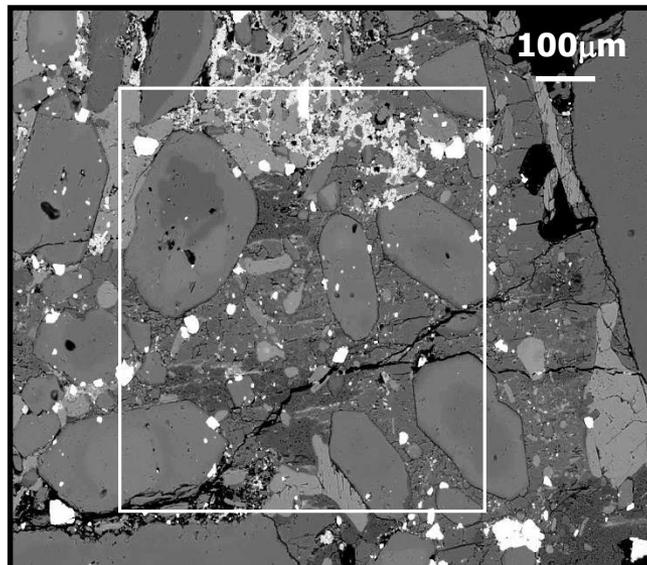
Prerequisite: kimberlites from Udachnaya-East (Siberia), EKATI cluster, Gahcho Kué, Diavik, Jericho (Canada), Majuagaa (W. Greenland), Wesselton, Bultfontein, Venetia (S. Africa), pipe #1 (Finland) and Antarctica.



Udachnaya-East : an unaltered “flagship” in kimberlite family

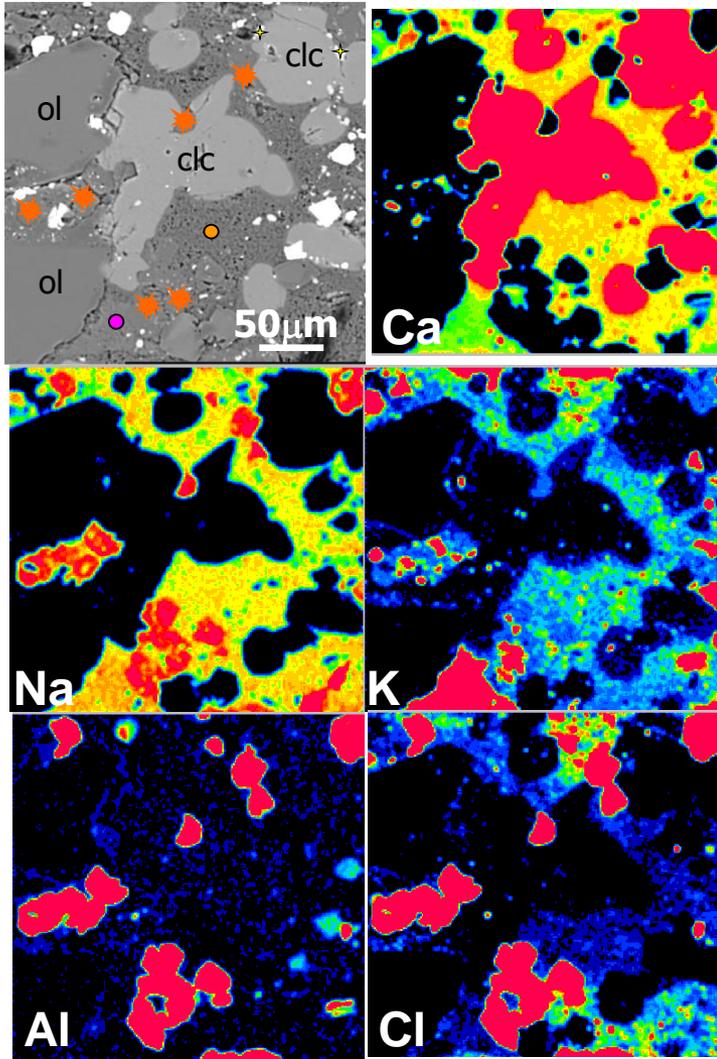


	wt%
SiO ₂	26.71
TiO ₂	1.25
Al ₂ O ₃	1.75
FeO	8.09
MnO	0.14
MgO	31.33
CaO	12.19
Na ₂ O	3.23 <0.4
K ₂ O	1.33
P ₂ O ₅	0.49
SO ₃	0.48
CO ₂	9.42
H ₂ O	0.38 >2.7
Cl	2.38
Total	99.17
H ₂ O-leach	9.24

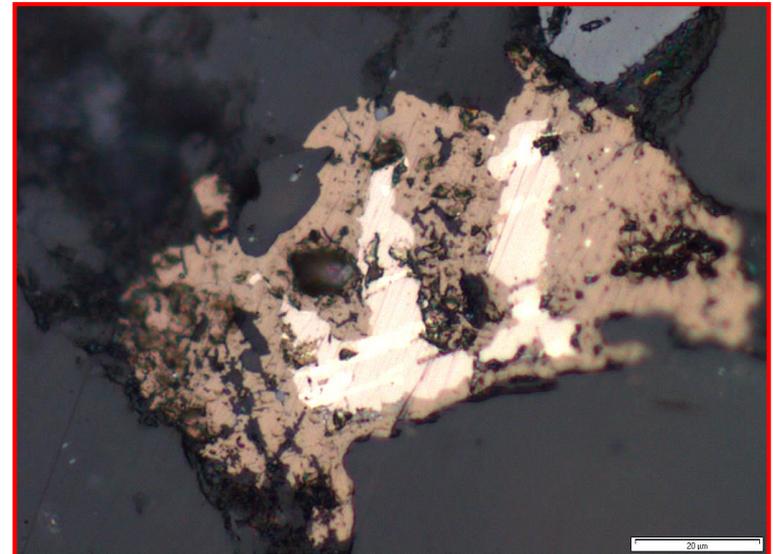
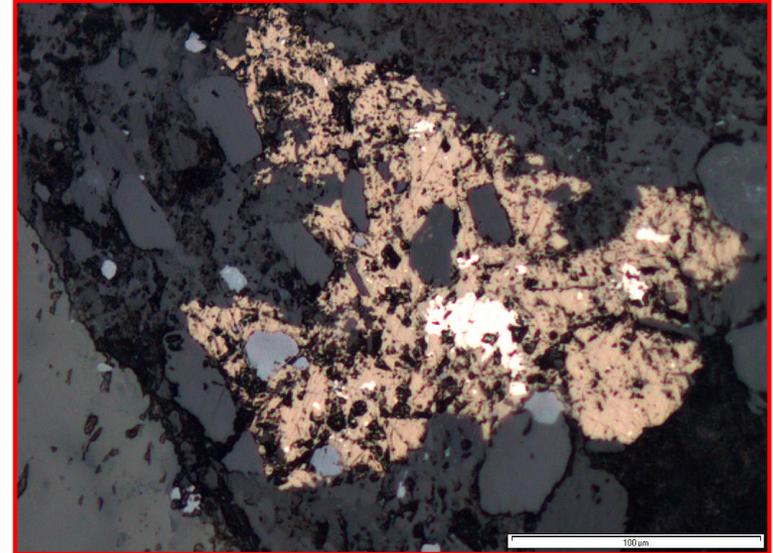


Udachnaya-East : an unaltered “flagship” in kimberlite family

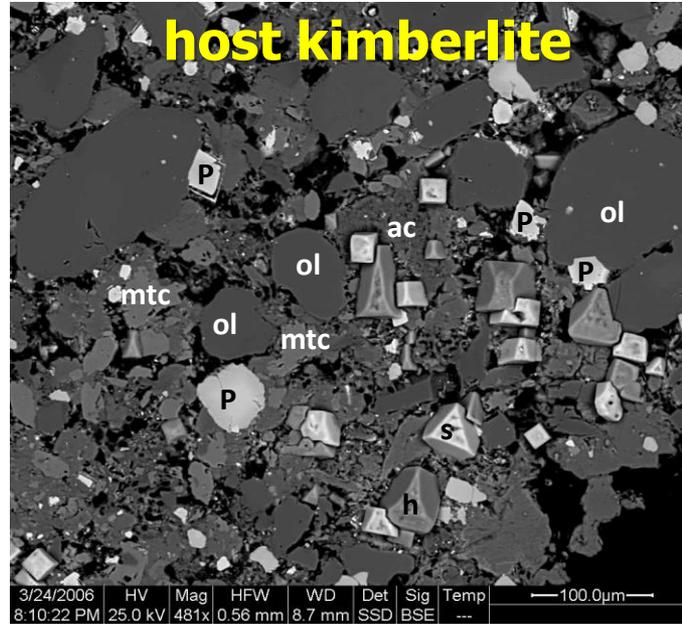
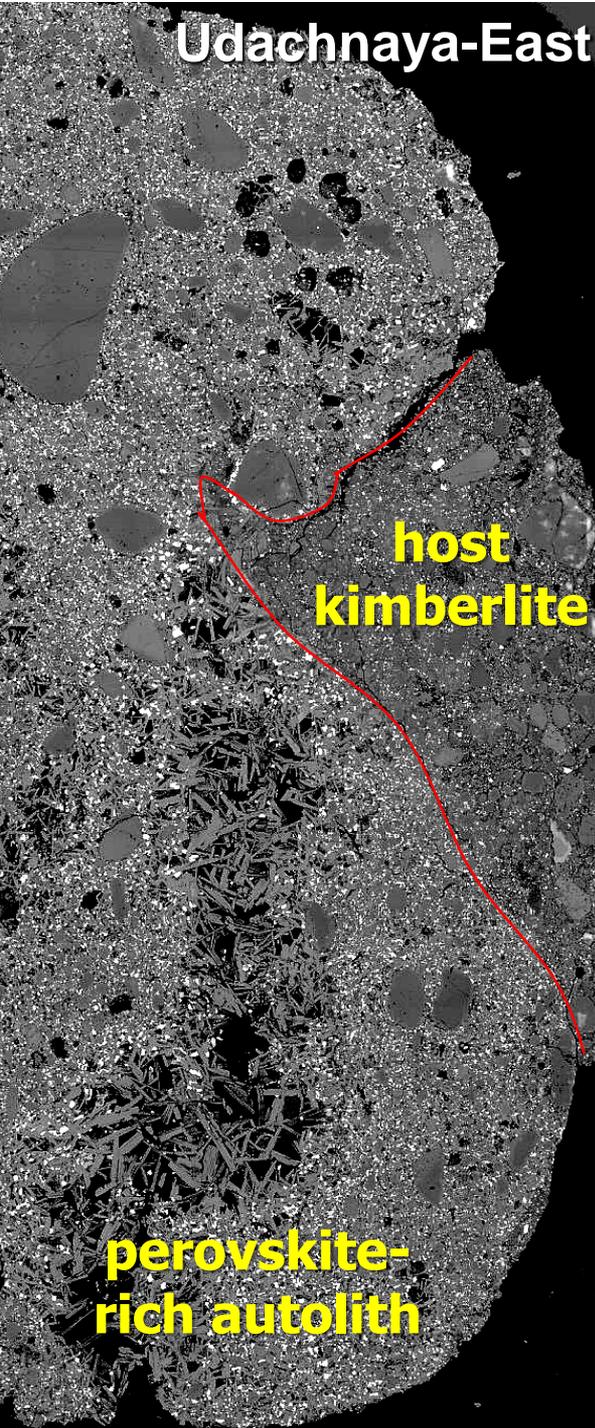
★ Sodalite $\text{Na}_8\text{Al}_6\text{Si}_6\text{O}_{24}\text{Cl}$



Djerfisherite $\text{K}_6(\text{Cu},\text{Fe},\text{Ni})_{25}\text{S}_{26}\text{Cl}$
Rasvumite KFe_2S_3



Udachnaya-East : an unaltered “flagship” in kimberlite family

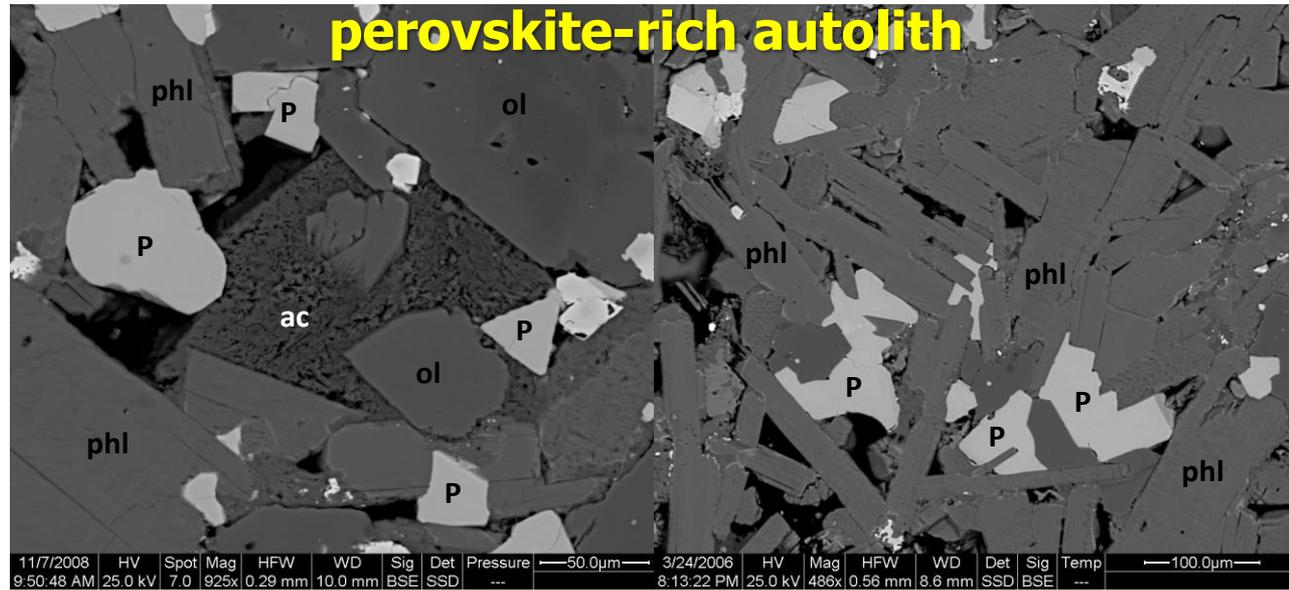


Mantle origin:
isotope compositions
(perovskite)

$$^{87}\text{Sr}/^{86}\text{Sr} = 0.7031$$

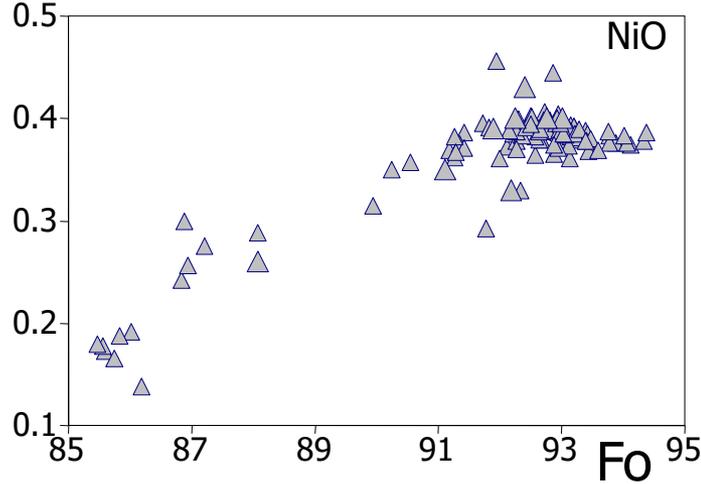
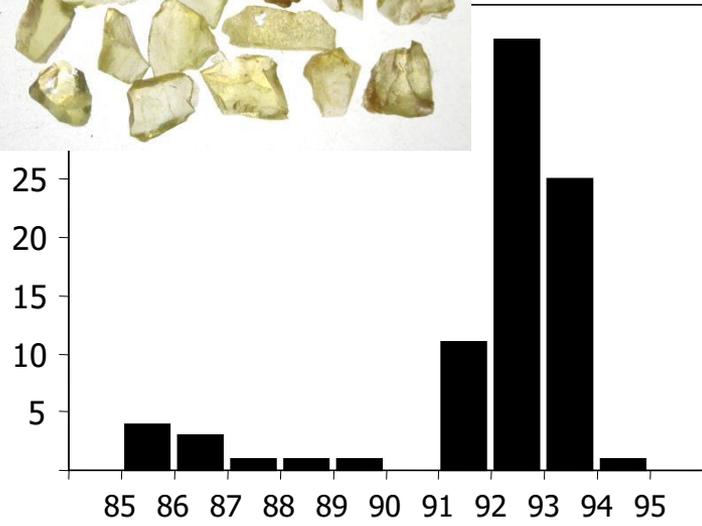
$$\epsilon\text{Nd} = +5.0$$

$$\epsilon\text{Hf} = +5.3$$

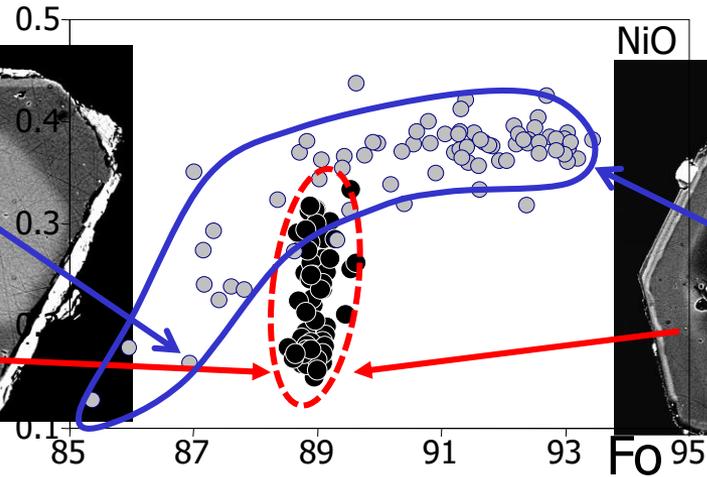
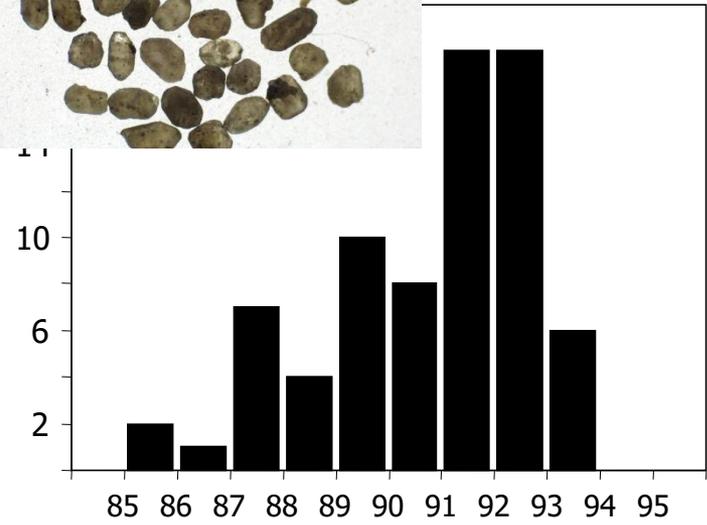


Udachnaya-East : an unaltered “flagship” in kimberlite family

Macrocrysts

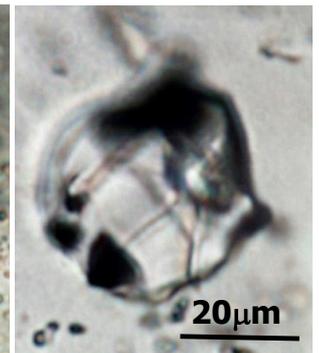
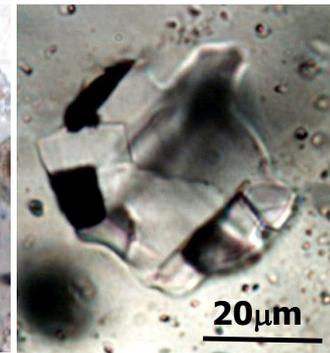
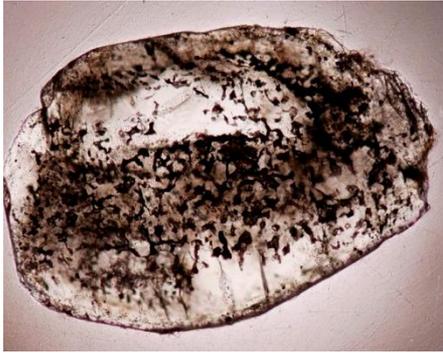


Phenocrysts



Alkali carbonate-chloride melt inclusions: common in ALL kimberlites

Udachnaya-East

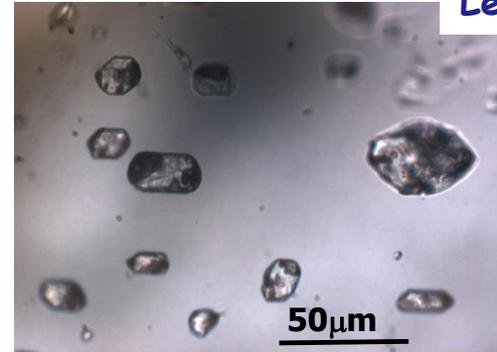


Jericho

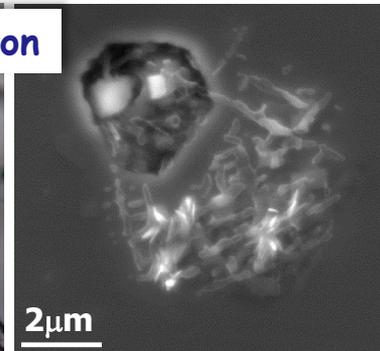
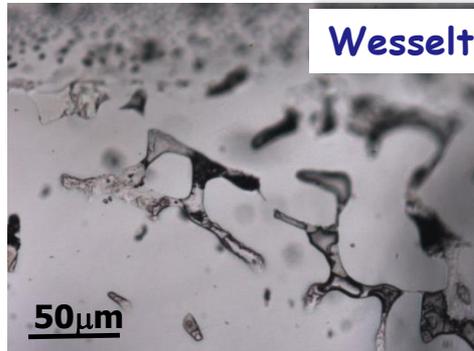
220C



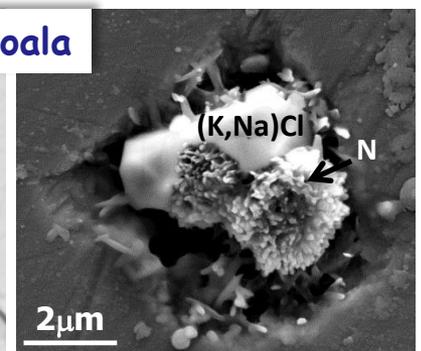
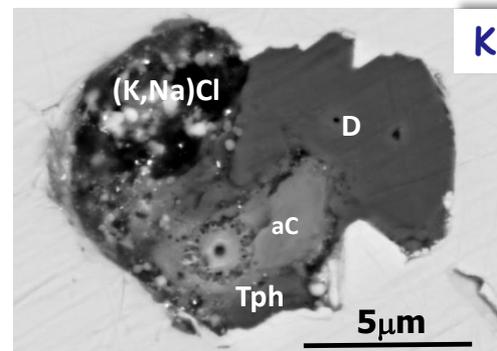
Leslie



Wesselton

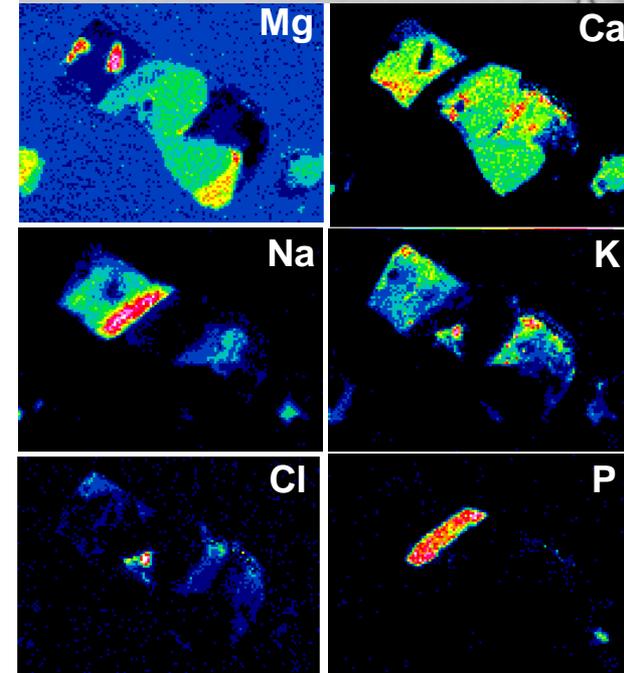
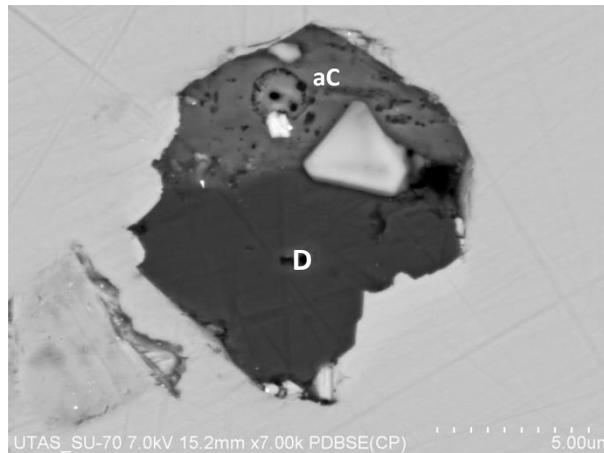
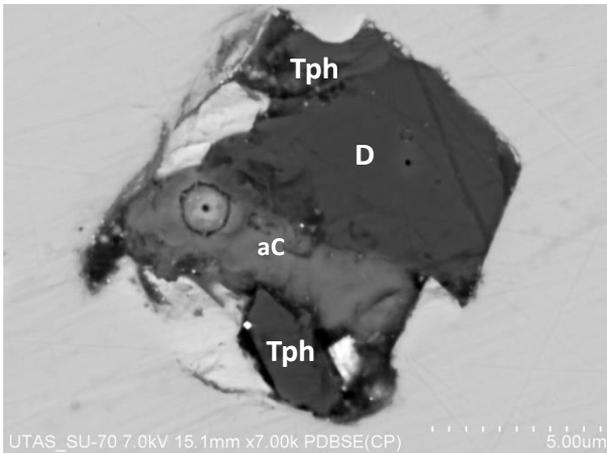
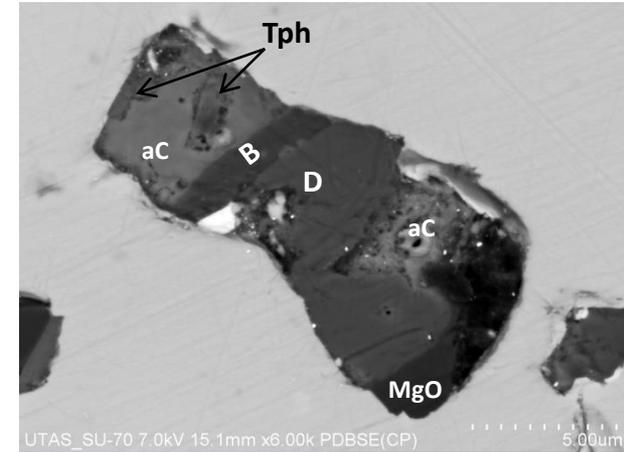
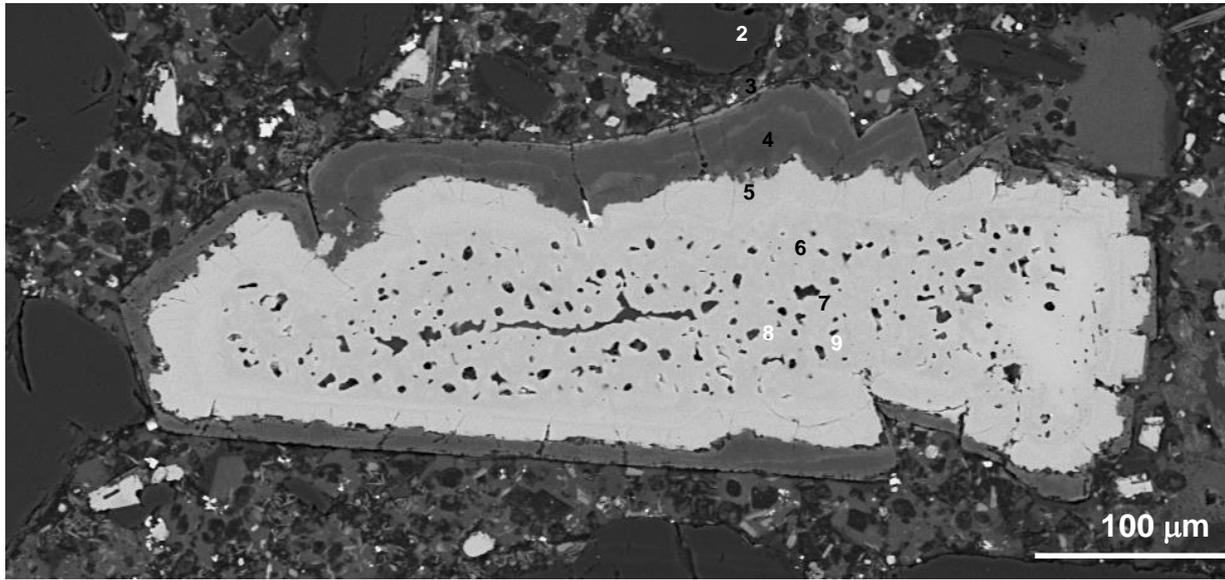


Koala



Alkali carbonate-chloride melt inclusions: common in ALL kimberlites

Cr-spinel, Koala pipe, EKATI cluster, Canada

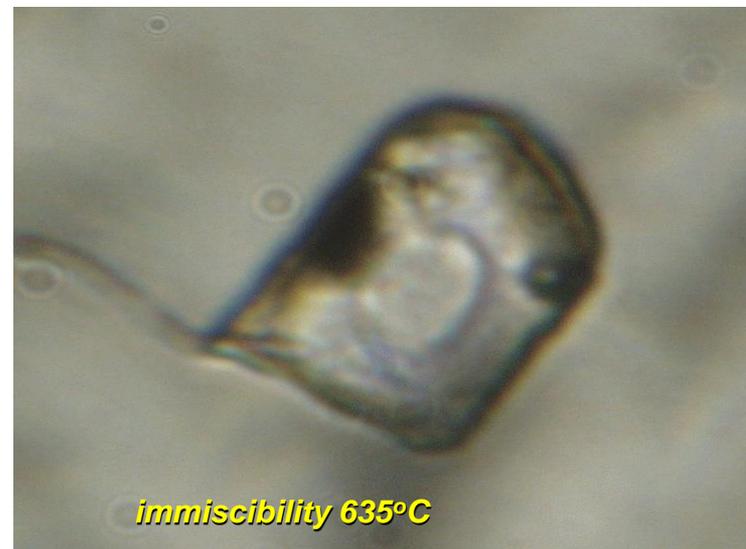


Melt homogenisation temperature ~ 670°C;

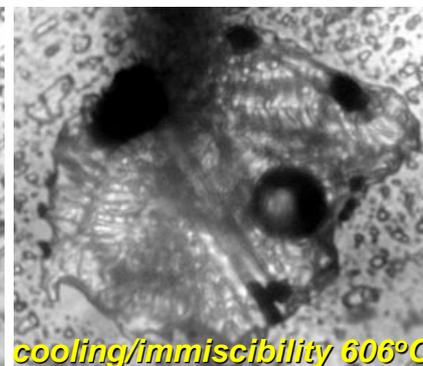
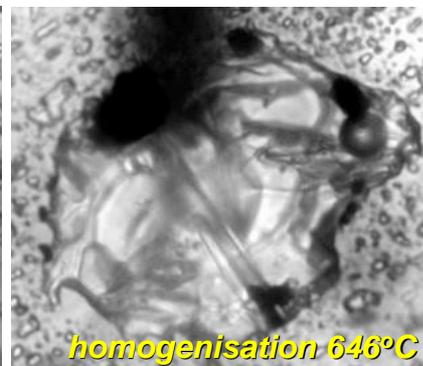
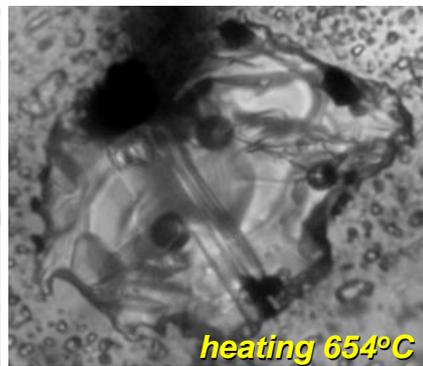
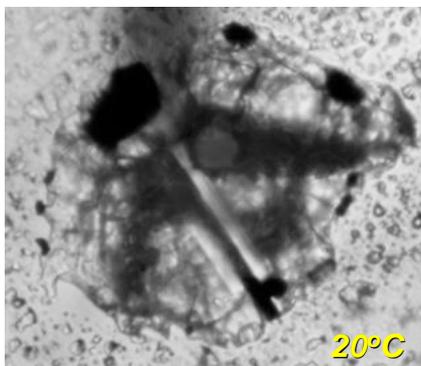
Carbonate-chloride melt immiscibility < 670°C:

common in ALL kimberlites

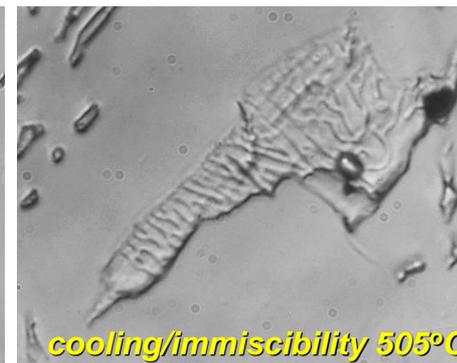
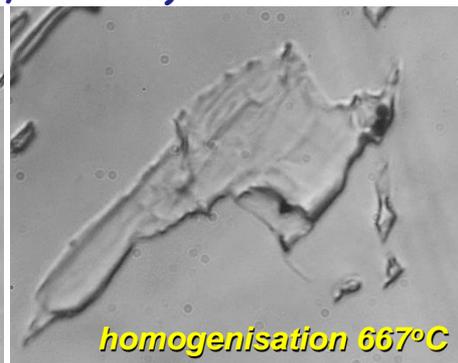
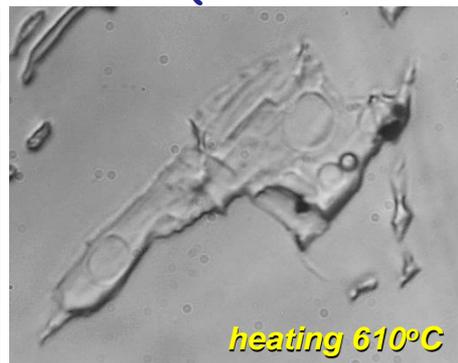
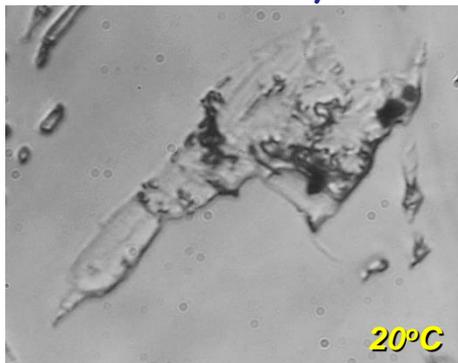
Olivine, Leslie kimberlite
(EKATI cluster, Canada)



Olivine, Udachnaya-East kimberlite (Siberia, Russia)



Olivine, Koala kimberlite (EKATI cluster, Canada)



**Melt homogenisation temperature ~ 670°C:
common in ALL kimberlites**



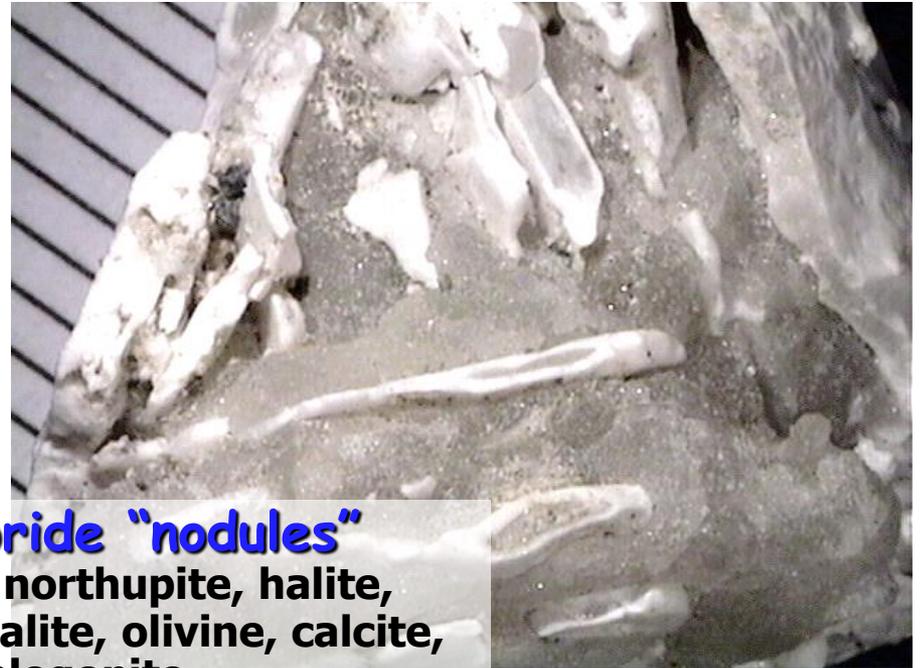
**Carbonate-chloride melt immiscibility < 670°C:
common in ALL kimberlites**

Udachnaya-East kimberlite: carbonate-chloride “pools” of residual melt

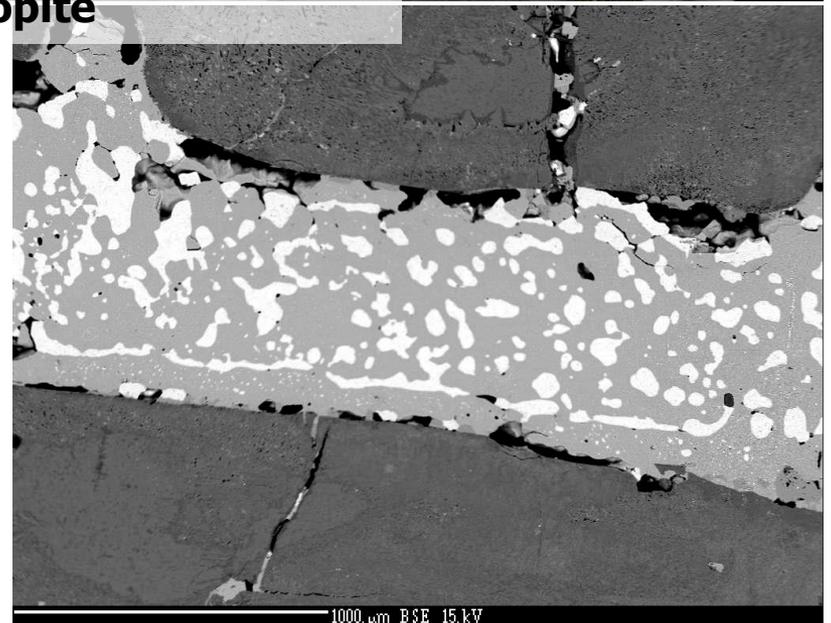


Chloride “nodules”

Udachnaya-East kimberlite: carbonate-chloride “pools” of residual melt



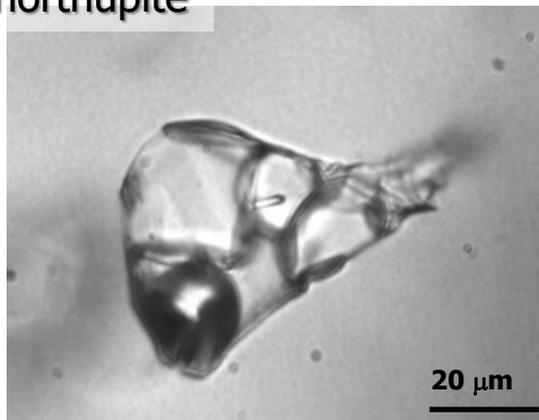
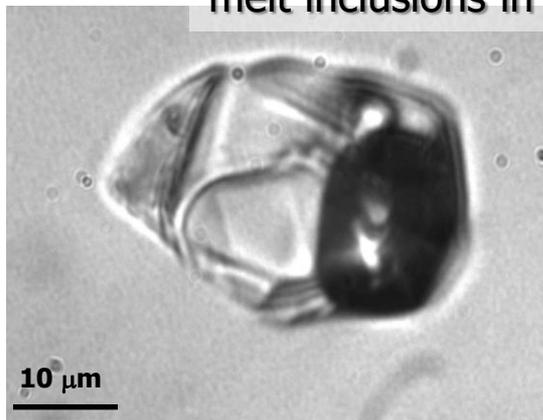
Carbonate- chloride “nodules”
nyerereite; shortite; northupite, halite,
sylvite, apatite, apthitalite, olivine, calcite,
tetraferriphlogopite



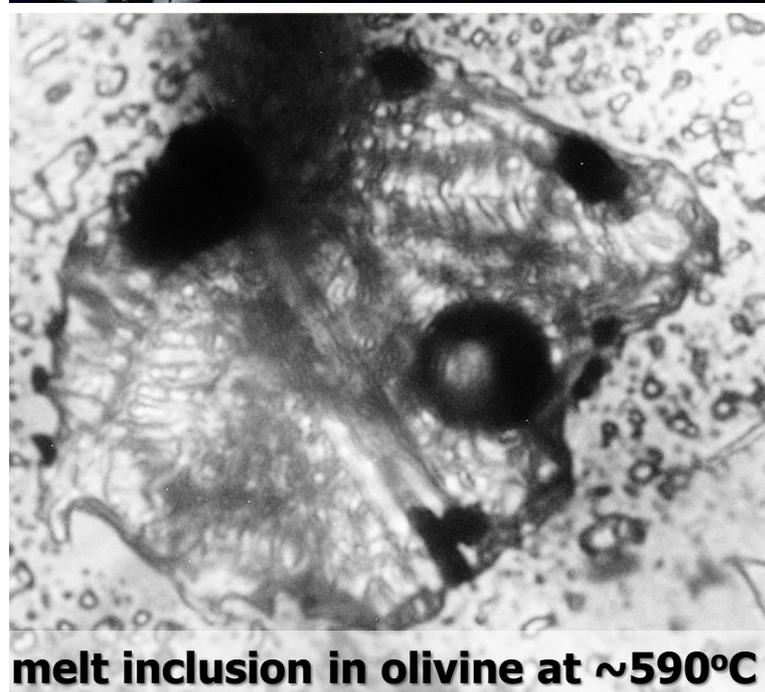
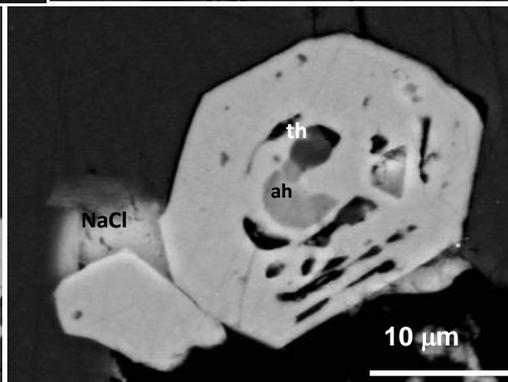
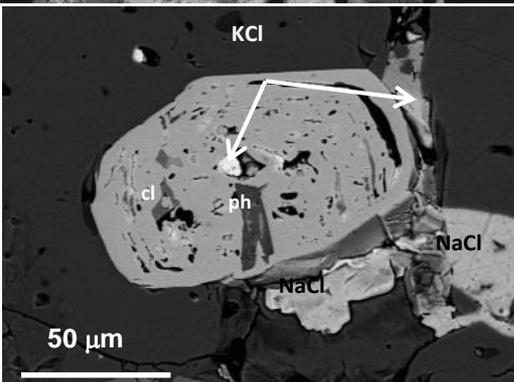
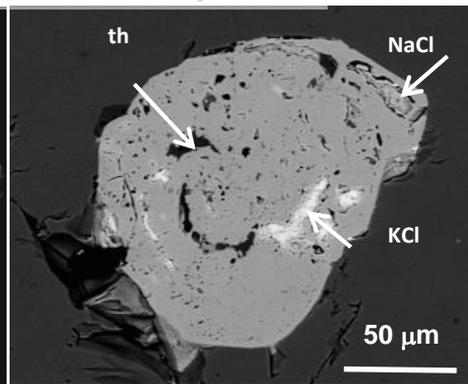
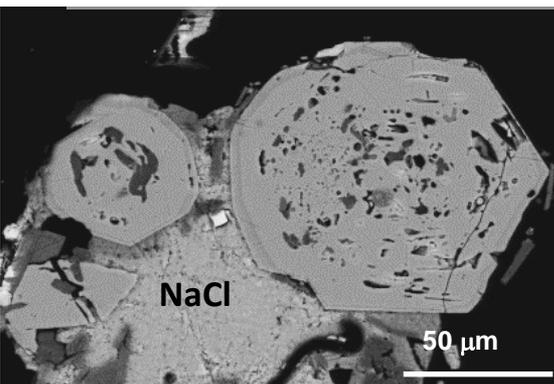
1000 μm BSE 15 kV

Udachnaya-East kimberlite: carbonate-chloride “pools” of residual melt

melt inclusions in northupite

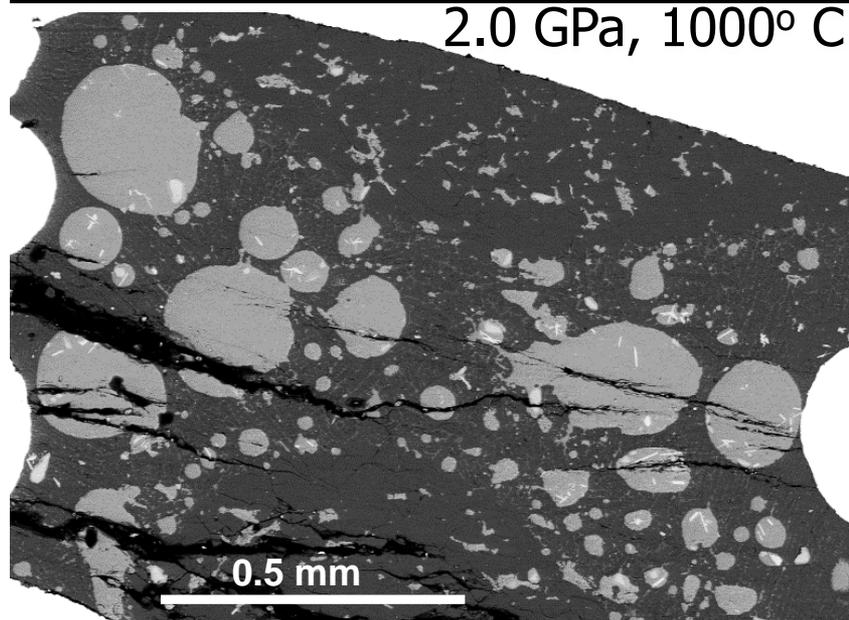
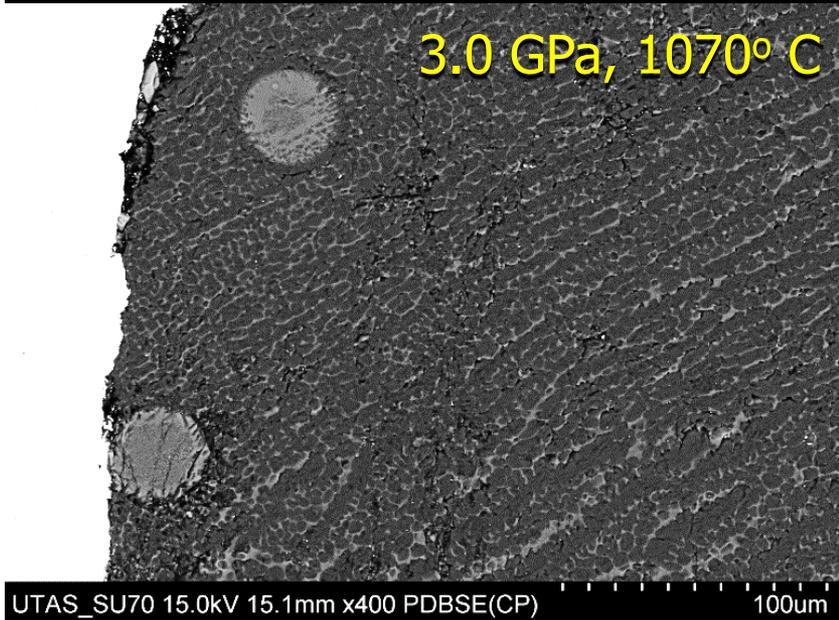
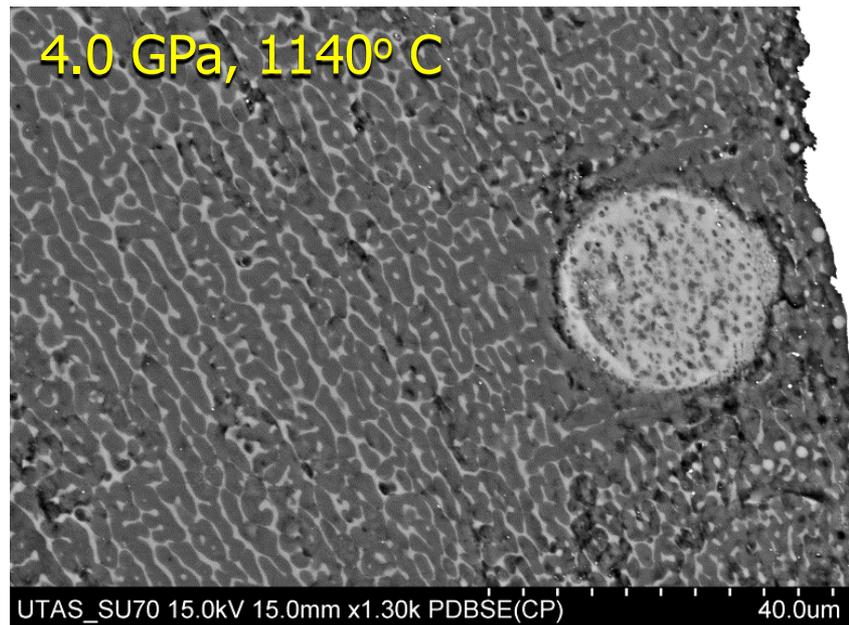


melt inclusions in shortite-hosted apatite



melt inclusion in olivine at ~590°C

Carbonate-silicate melt immiscibility in the system 80%Na₂CO₃ – 20%Opx at mantle conditions: experimental approach



New “kimberlite ascent” model: a storyline



SCLM

What drives explosive emplacement of kimberlites?

- 1. Alkali carbonate - chloride primary melt** composition can be responsible for inferred rheological properties (low density and viscosity and related high buoyancy) and fast ascent to the surface and preservation of diamonds;
- 2. Kimberlite primary melt** entraps mantle and crustal rocks, dissolves mantle silicate minerals (essentially orthopyroxene), becomes S-saturated, experiences carbonate-silicate liquid unmixing and becomes loaded and saturated in **olivine**;
- 3. Neither primary nor secondary water** is present, as manifested by rock composition, fresh olivine and absence of serpentine;
- 4. No decarbonation** is caused by sub-surface crystallisation of olivine (< 10-20%) and melt decompression, so CO_2 is not released as a gaseous phase.

If H_2O and CO_2 gases are not involved
what causes explosive eruption?

Kimberlite Magma emplacement: new facts

Kimberlite explosions are unexpectedly **powerful** for such small volumes

"The initial breakthrough of the magma at the surface is likely to have been violent and highly explosive...with excavation of the pipe from the top down, as the fragmentation surface in the magma column drops quickly upon vent opening"
(**Porritt & Cas, 2009**)

"At Lac de Gras...excavated steep-sided pipes were empty to ~400-500 m depth before eventually being infilled by kimberlite"
(**Kopylova & Hayman, 2008**)

"The sustained eruption proceeds to "drill" down into the country rock, efficiently excavating down to >750 m below surface" (**Moss et al., 2009**)

"...the (Venetia) K2 eruption was a protracted explosive event ...presence of layered breccias and pyroclastic rocks to a considerable depth (>900 m) requires that the pipe was deeply excavated prior to filling" (**Brown et al., 2009**)

Explosions occur when kimberlite is **cold and solid**

"From a few of the diamond pipes rare leaves, turtle bones and fish parts have been recovered....Abundant wood specimensare in various states of carbonization. Some specimens have not been exposed to any burning and look like normal cedar logs"
www.napegg.nt.ca/newsletters/oct99

"...Relatively fresh, often charred – but not petrified – wood found only in China today, has been encountered in drill core at depths up to 400 metres. ...coniferous trees were uprooted and incorporated into the pipes"
<http://www.diavik.ca/Geo.htm>

"Most clasts...show no evidence of being thermally metamorphosed. The presence of bituminous shale and carbonized wood indicate that high temperatures did not occur during diatreme emplacement " (**Mitchell, 1986**)

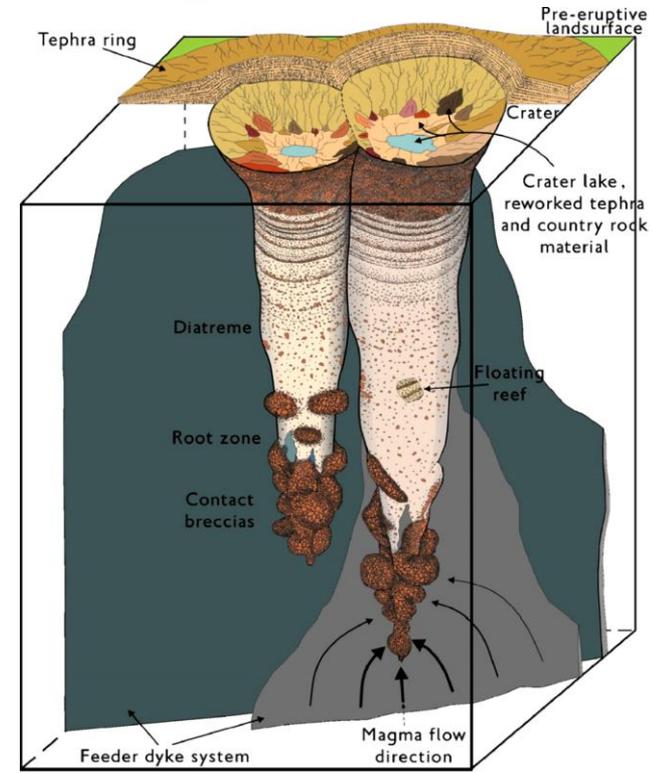


Uncharred wood in Diavik kimberlite
(photo Greg Yaxley)

Kimberlite emplacement: prevailing models

1. Fluidisation: rapid degassing and expansion of magmatic volatiles (CO_2 and H_2O) in an open system that creates "...a bed of particles, which develops fluid-like properties as a consequence of the flow of interstitial gas" (Sparks et al., 2006).

2. Hydrovolcanism/phreato-magmatism: interaction of magma or magmatic heat with an external source of water... "diatremes form on dykes in hard rocks ..wherever there is a local potential for thermohydraulic explosive interaction of the rising magma with groundwater contained in fractures and faults" (Lorenz & Kurszlaukis, 2007).



V. Lorenz & S. Kurszlaukis, 2007

"Emplacement of the (Yubileinaya) pipe occurred over an **extended time span** with intermittent phases of volcanic quiescence and consolidation.... There is neither textural evidence that **violent degassing of a juvenile gas phase** has caused pipe excavation, nor that **external water** was present during the main phase of volcanic eruptions" (Kurszlaukis et al., 2009, Lithos)

Kimberlite emplacement: new model

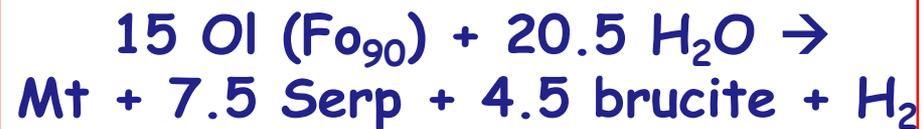
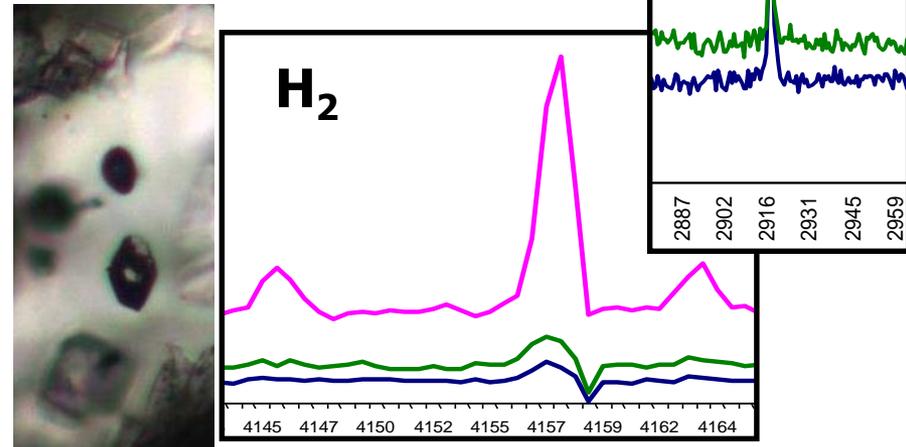
Magma-derived and serpentisation-derived **CH_4** and **H_2** blended **with air** result in combustion and powerful explosion

New kimberlite emplacement model: Rationale

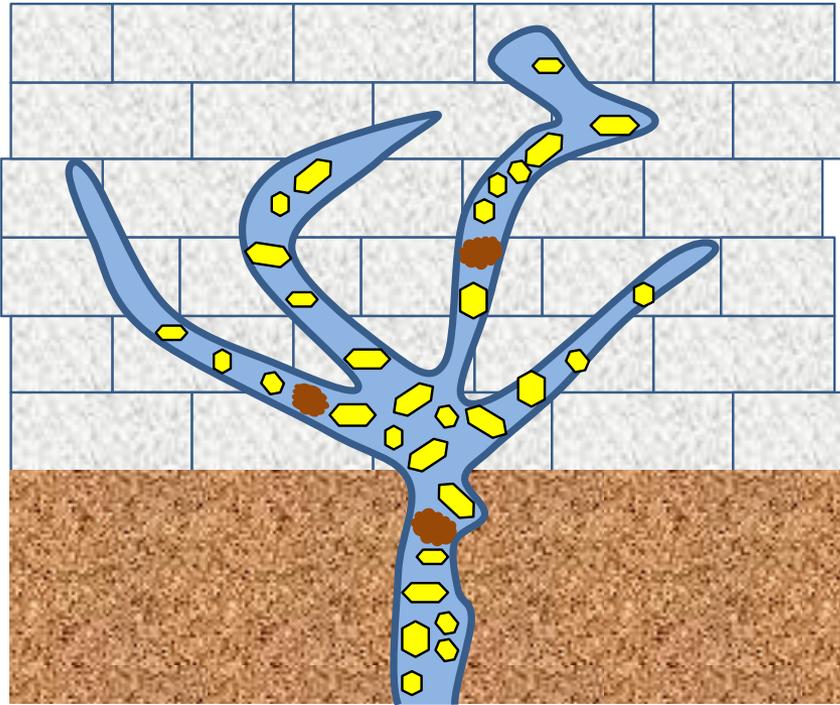
1. Spontaneous outgassing: $\sim 10^5$ m³/day of gases N₂, CH₄ and H₂ at 50-70 atm from deep boreholes drilled through the unaltered block of the Udachnaya-East kimberlite (Marshintsev, 1986; Drozdov et al., 1989)

2. CH₄ and H₂ in olivine-hosted fluid inclusions in the Gahcho Kué kimberlite (Slave craton, Canada):
Raman spectroscopy

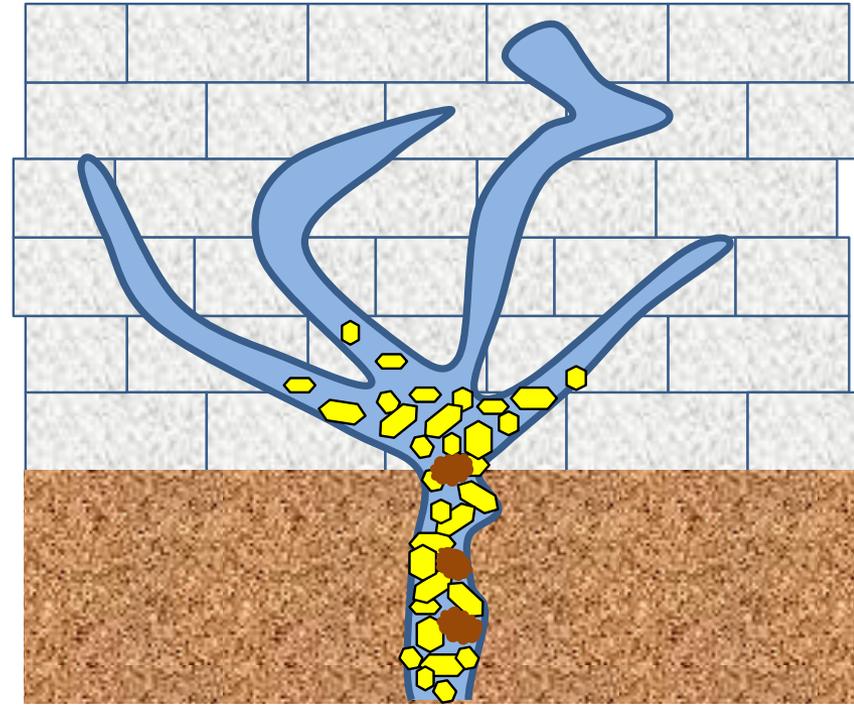
3. CH₄ and H₂ are produced during serpentinisation of olivine: examples from present-day CH₄ and H₂ production in various continental (Abrajano et al., 1988; Neal and Stanger, 1983; Coveney et al., 1987; Charlou et al., 1998) and oceanic areas (Haggerty, 1991; Kelley, 1996) where mafic-ultramafic rocks are present.



New kimberlite emplacement model: a storyline

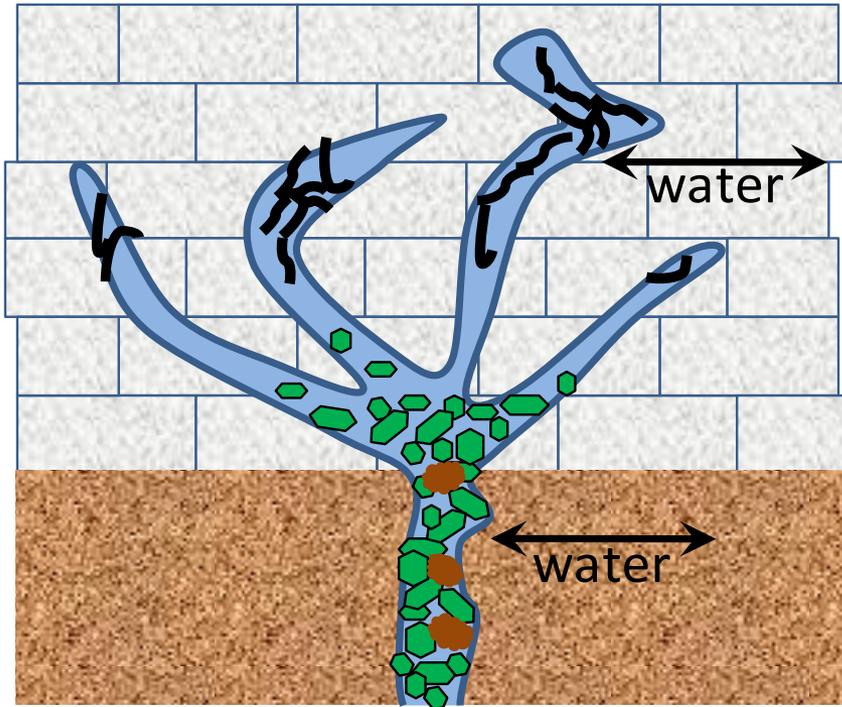


1. Emplacement of the kimberlite magma (~65% carbonate-chloride-silicate liquid + ~35% solids – olivine, mantle & crustal xenoliths and xenocrysts) as subsurface dykes

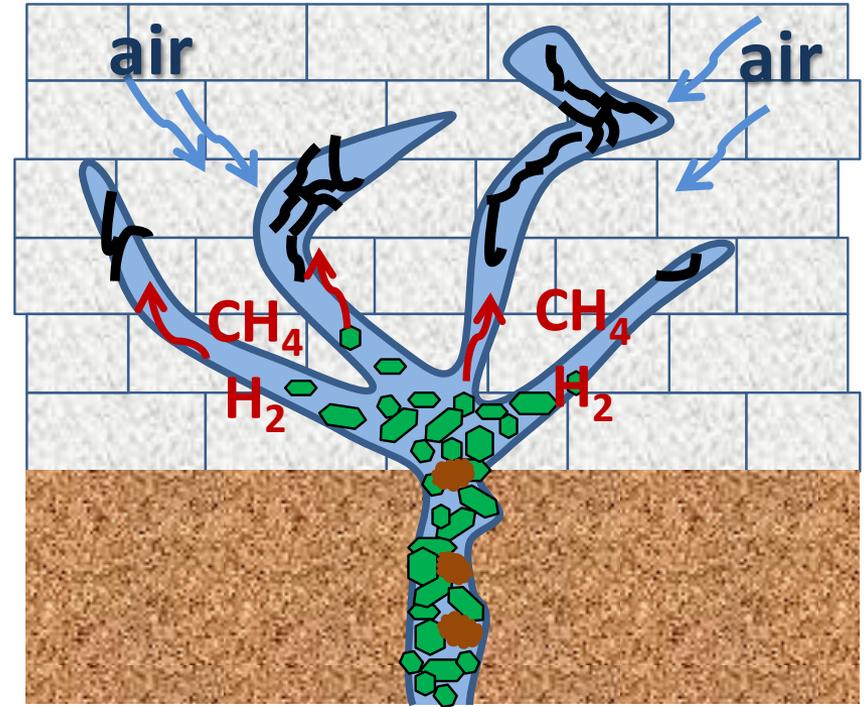


2. Gravitational separation of solids from the residual carbonate-chloride liquid in dykes. Olivine-rich cumulates with interstitial melt ("root zones" of hypabyssal kimberlites). Solidifying melt-rich zones with scattered solids in the upper parts of dykes.

New kimberlite emplacement model: a storyline

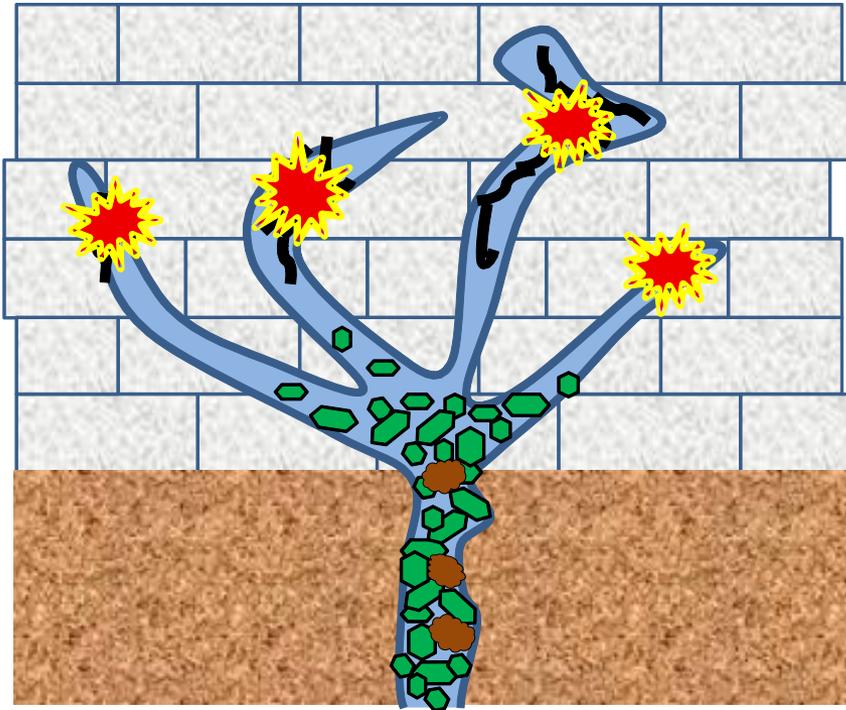


3. Water-soluble alkali carbonates and chlorides are leached out. Interconnected caverns are created throughout upper parts of dykes; whereas water causes serpentinisation of olivine in the “root zones”

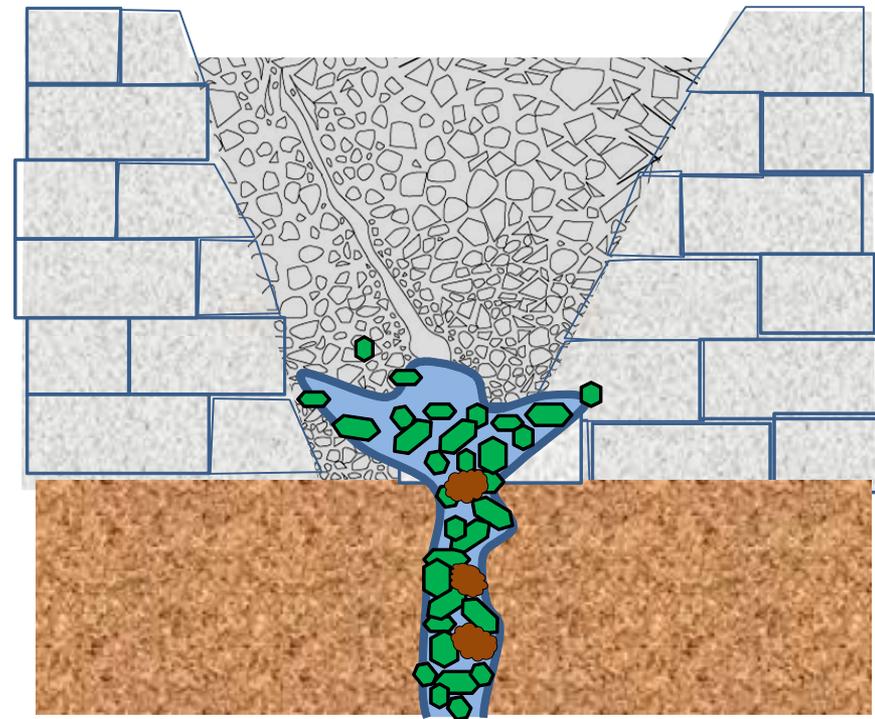


4. Cavernous system act as a “vacuum pump” for free gases in this environment: magmatic, postmagmatic and ambient gases (mixture of air, water vapour and magma-derived volatiles).

New kimberlite emplacement model: a storyline



5. Gas-mixing and pressurisation (by expanding serpentine) in caverns cause detonation. One explosion may cause chain reaction by sending shock waves.



6. Vertical and lateral explosive boring. Fragmentation inside the dyke system and country rocks . Collapse of rocks from the top and walls  Growth of carrot- shaped “diatremes” by excavation from top down. Natural autoclave for microdiamonds.

New “kimberlite emplacement” model: a storyline





Udachnaya (*fortuitous*) Удачная model of kimberlite

1. new approach to kimberlite melt composition and temperature
2. new estimates of rheological properties of kimberlite magmas
3. new mechanism of the kimberlite magma ascent
4. new ideas on the source of explosive energy/forces